

# The use of mass spectrometry in lipidomics

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## Outlines

Brief introduction to lipidomics

Analytical methodology:  
MS/MS structure elucidation  
of phospholipids

Phospholipid analysis in  
lean and ob/ob mice by  
mass spectrometry

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**Lipids are  
very  
important!!**

**Structural components  
of cell membrane**

**Nutrition**

**Energy source/storage**

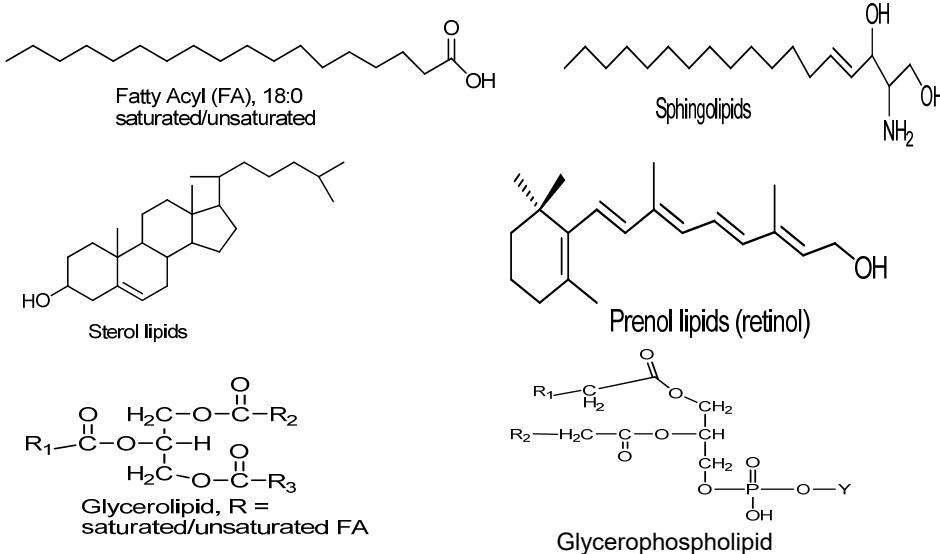
**Signaling molecules  
(extra- and intra-cellular)**

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- Lipidomics- A comprehensive analysis of lipid molecules in response to cellular stress and challenges
- Changes in lipid interactions

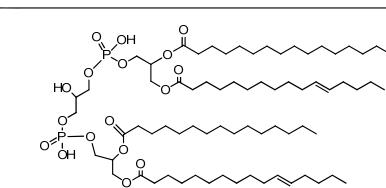
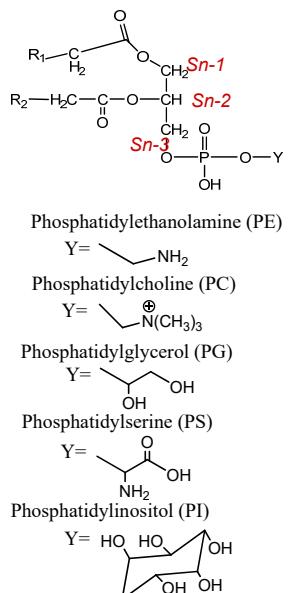
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## Structures of different lipids classes



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## Structures of main phospholipids



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## Extraction of lipids by Bligh/Dyer method

- To a homogenized sample (1 ml containing internal standards) add methanol (2.5 ml) and chloroform (1.25 ml), sonicate by 4-5 bursts and added 1.0 ml water and 1.25 ml chloroform additionally and vigorously shaken.
- Centrifuge (1,000 x g) for 2 min and separate the chloroform layer (bottom layer) and repeat the process twice.
- Combine the chloroform soluble phase and evaporate to dryness and stored at -20 °C until analysis.

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## Shotgun lipidomics: intrasource separation of lipids for quantitative lipidomics

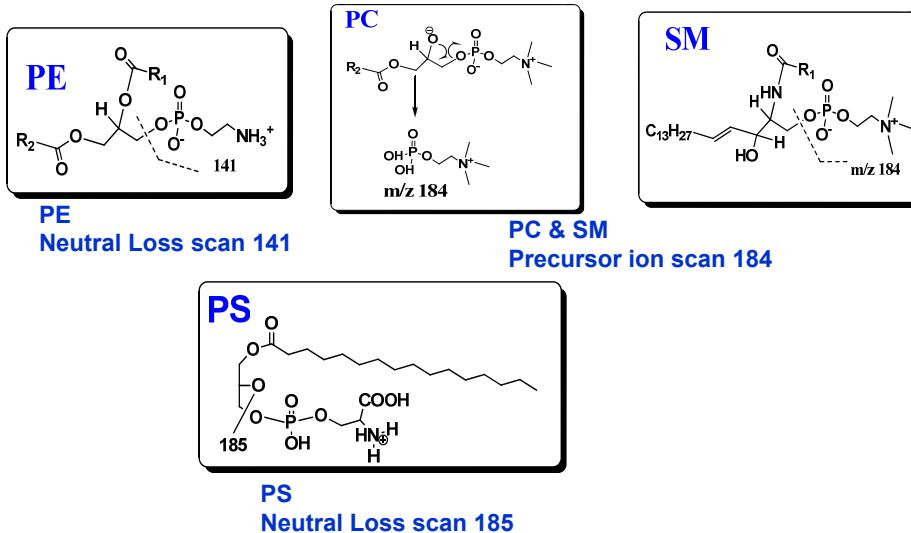
Group	Electrical Propensity	Lipid Classes
Anionic lipids	Carry net negative charge(s) at physiological pH	Cardiolipin, acylCoA, sulfate, PtdIns (PtdInsP, PtdInsP <sub>2</sub> , PtdInsP <sub>3</sub> ), PtdGro, PtdSer, PtdH, etc.
Weak anionic lipids	Carry a net negative charge at alkaline pH	PE, lysoPE, ceramide, NEFA, eicosanoids, etc.
Neutral polar lipids	Neutral at alkaline pH	PC, lysoPC, SM, glycolipid, TAG, etc.
Special lipids	Vary	Acylcarnitine, sterols, etc.

The ionization efficiency of an analyte greatly depends on the electrical propensity of an individual analyte in its own microenvironment to lose or gain a charge

Source: Gross and Han, Mass Spec Rev. 2005

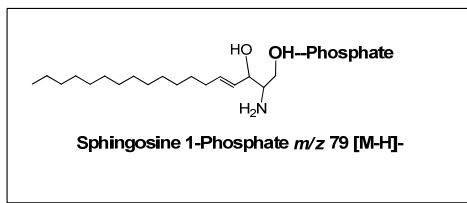
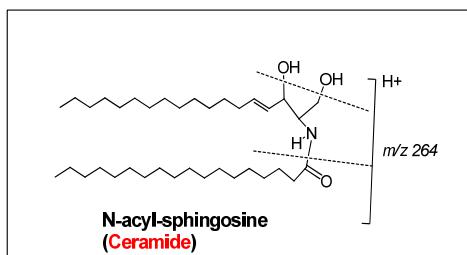
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## Profiling glycerophospholipids in a complex mixture using MS/MS



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## How to profile sphingolipids in a complex mixture using MS/MS?

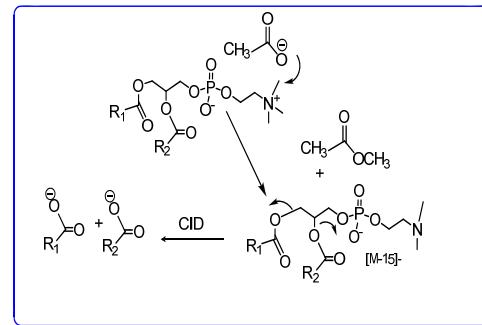


$m/z$  264 is a characteristic ion for all compounds containing a sphingosine backbone in +ve ion mode

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## Phosphatidylcholine loses a methyl group to form a negatively charged, pseudomolecular ion

- Phospholipids may undergo demethylation and then the loss of the fatty acyl groups from glycerophosphocholine backbone.



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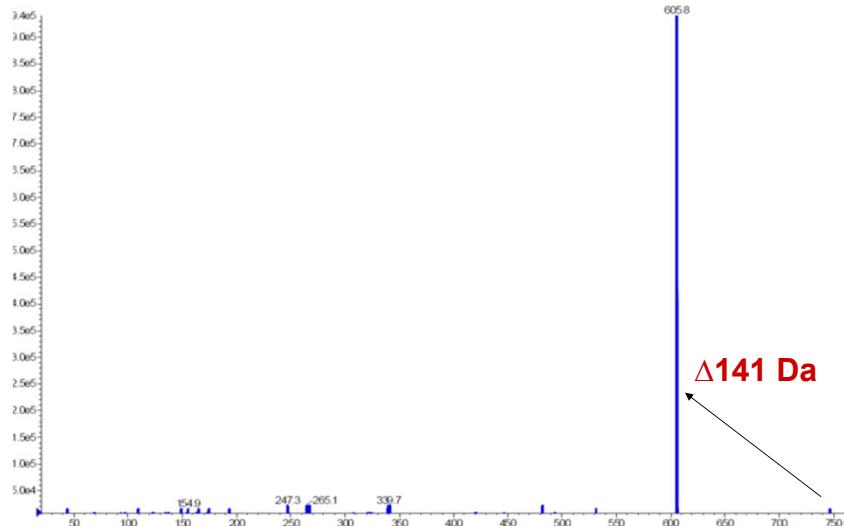
### ESI-MS/MS analyses of various lipids

Lipid Class(s)	Precursor Ion	MS/MS Mode & Conditions	Fragment
cardiolipin	[M-2H] <sup>2-</sup>	PI, m/z 153.0, 35 eV	glycerol phosphate derivative
PtdGro, PtdH	[M-H] <sup>-</sup>	PI, m/z 153.0, 35 eV, *	glycerol phosphate derivative
PtdIns	[M-H] <sup>-</sup>	PI, m/z 241.1, 45 eV	cyclic inositol phosphate
		PI, m/z 153.0, 35 eV	glycerol phosphate derivative
PtdInsP	[M-H] <sup>-</sup>	PI, m/z 321.1, 53 eV	phosphoinositol phosphate
PtdInsP <sub>2</sub>	[M-H] <sup>-</sup>	PI, m/z 401.1, 62 eV	diphosphoinositol phosphate
PtdSer	[M-H] <sup>-</sup>	NL, 87.0 amu, 25 eV, *	serine
		PI, m/z 153.0, 35 eV	glycerol phosphate derivative
sulfatide	[M-H] <sup>-</sup>	PI, m/z 97.0, 65 eV	sulfate
acylCoA	[M-2H] <sup>2-</sup>	PI, m/z 339.0, 30 eV, *	doubly-charged CoA derivative
PE, lysoPE	[M-H] <sup>-</sup>	PI, m/z 196.0, 50 eV	glycerol phosphoethanolamine derivative
ceramide	[M-H] <sup>-</sup>	NL, 256.2 amu, 32 eV *	
		NL, 327.3 amu, 32 eV	
		NL, 240.2 amu, 32 eV *	2-trans-palmitoyl alcohol
PC, lysoPC, SM	[M+Li(Na)] <sup>+</sup>	NL, 59.1 amu, -28 eV, *	trimethylamine
	[M+Li(Na)] <sup>+</sup>	NL, 183.1 amu, -32 eV	phosphocholine
	[M+Li] <sup>+</sup>	NL, 189.1 amu, -42 eV	lithium cholinephosphate
	[M+Na] <sup>+</sup>	NL, 205.1 amu, -35 eV	sodium cholinephosphate
	[M+H] <sup>+</sup>	PI, m/z 184.1, -30 eV, *	phosphocholine
	[M+Cl] <sup>-</sup>	NL, 50.0 amu, 24 eV, *	methylchloride
cerebroside	[M+Li] <sup>+</sup>	NL, 162.2, -50 eV, *	
	[M+Cl] <sup>-</sup>	NL, 36.0 amu, 30 eV	hydrogen chloride
MGDG	[M+Li(Na)] <sup>+</sup>	PI, m/z 227(243), -45 eV	Li(Na)+galactose derivative
DGDG	[M+Li(Na)] <sup>+</sup>	PI, m/z 227(243), -66 eV	Li(Na)+galactose derivative
acylcarnitine	[M+H] <sup>+</sup>	PI, m/z 85.1, -20 eV, *	carnitine
chol. ester	[M+NHa] <sup>+</sup>	PI, m/z 369.3, -50 eV, *	cholestane cation
TAG	[M+Li] <sup>+</sup>	NL, X amu, -35 eV	a fatty acid

Source: Gross and Han, Mass Spec Rev. 2005

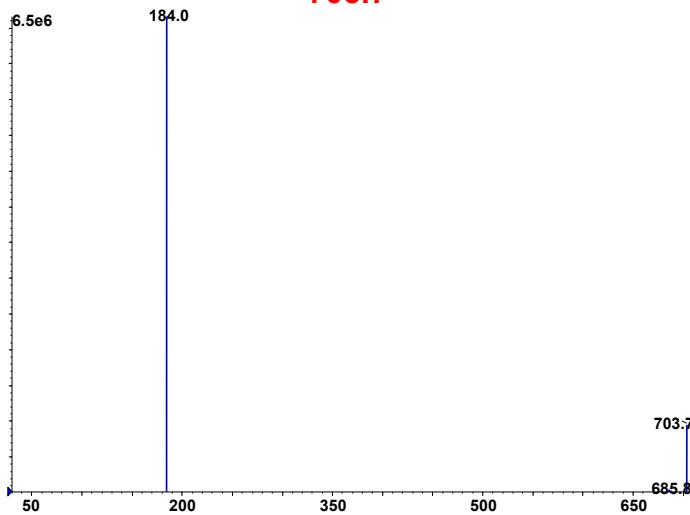
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**MS/MS of m/z 746.90: PE Std**  
**Neutral loss of 141 is a characteristic for detecting PE**



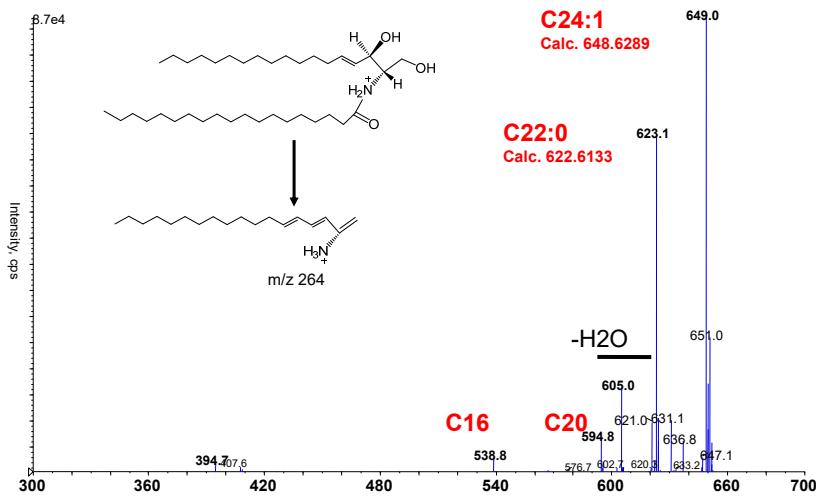
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**MS/MS of sphingomyelin standard (2S,3R,4E)-  
2-acylaminoctadec-4-ene-3-hydroxy-1-Phosphocholine m/z  
703.7**



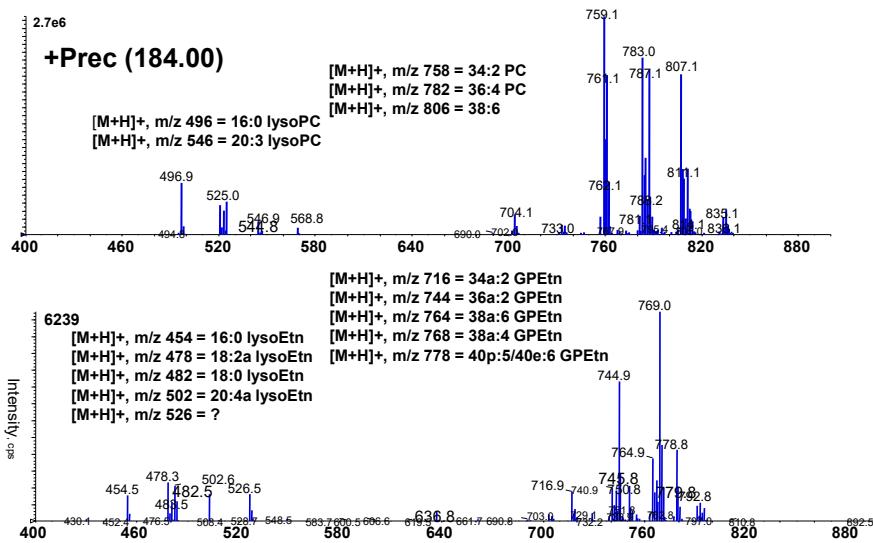
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## Precursor ion scan m/z 264 in +ve ion mode is specific to identify ceramides in a sample



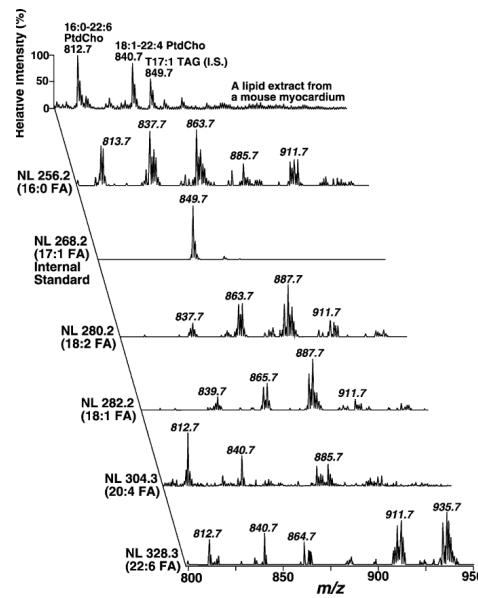
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## Profiling of phospholipids using precursor ion m/z 184 and neutral loss scan 141 for PC, SM and PE



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**Neutral loss scans can be used to profile triacylglycerides (TAG)**



Source: Gross and Han, Mass Spec Rev. 2005

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**Library search for eicosanoid**  
<http://www.lipidmaps.org/>

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**LIPID Metabolites And Pathways Strategy**

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**LMSD: Lipid classification search results**

Uni-ID	Common Name	Systematic Name	Formula	Mass
LIPAC0000001	8(9)-EpETE	(+/-)-8(9)-epoxy-5Z,11Z,14Z-icosatetraenoic acid	C <sub>20</sub> H <sub>32</sub> O <sub>2</sub>	318.22
LIPAC0000002	11(12)-EpETE	(+/-)-11(12)-epoxy-5Z,8Z,14Z-icosatetraenoic acid	C <sub>20</sub> H <sub>32</sub> O <sub>2</sub>	318.22
LIPAC0000003	14(15)-EpETE	(+/-)-14(15)-epoxy-5Z,8Z,11Z,12Z-icosatetraenoic acid	C <sub>20</sub> H <sub>32</sub> O <sub>2</sub>	318.22
LIPAC0000004	17(18)-EpETE	(+/-)-17(18)-epoxy-5Z,8Z,11Z,14Z-icosatetraenoic acid	C <sub>20</sub> H <sub>32</sub> O <sub>2</sub>	318.22
LIPAC0000005	11(R)-HEDE	11R-hydroxy-12E,14Z-eicosadienoic acid	C <sub>20</sub> H <sub>32</sub> O <sub>2</sub>	324.27
LIPAC0000006	17R,18S-EpETE	17R,18S-epoxy-5Z,8Z,11Z,14Z-icosatetraenoic acid	C <sub>20</sub> H <sub>32</sub> O <sub>2</sub>	318.22
LIPAC0000007	15(R)-HEDE	15R-hydroxy-12E,14Z-eicosadienoic acid	C <sub>20</sub> H <sub>32</sub> O <sub>2</sub>	324.27
LIPAC0000008	15S-HEDE	15S-hydroxy-12E,14Z-eicosadienoic acid	C <sub>20</sub> H <sub>32</sub> O <sub>2</sub>	324.27
LIPAC0000009	Prostanoid acid skeleton	-	-	-
LIPAC0000010	6-keto-PGF <sub>1</sub> $\alpha$	6-oxo-9(10),15(16)-trihydroxy-13E-prostenoic acid	C <sub>20</sub> H <sub>32</sub> O <sub>5</sub>	370.24
LIPAC0000011	PGF <sub>2</sub> $\alpha$	9(10),15(16)-trihydroxy-5Z,13E-prostenoic acid	C <sub>20</sub> H <sub>32</sub> O <sub>5</sub>	354.24
LIPAC0000012	PGF <sub>2</sub> $\alpha$ (11)	9-oxo-11(12),15(16)-dihydroxy-5Z,13E-prostenoic acid	C <sub>20</sub> H <sub>32</sub> O <sub>5</sub>	352.22
LIPAC0000013	PGD <sub>2</sub> (11)	9,15(16)-dihydroxy-11-oxo-5Z,13E-prostenoic acid	C <sub>20</sub> H <sub>32</sub> O <sub>5</sub>	352.22
LIPAC0000014	PGA <sub>1</sub>	9-oxo-15(16)-hydroxy-10Z,13E-prostenoic acid	C <sub>20</sub> H <sub>32</sub> O <sub>5</sub>	336.23
LIPAC0000015	PGF <sub>2</sub> $\alpha$ -d4	9(10),15(16)-dihydroxy-5Z,13E-prostenoic acid (3,3,4,4-d4)	C <sub>20</sub> H <sub>32</sub> O <sub>5</sub>	350.27
LIPAC0000016	PGD <sub>2</sub> -d4	9,15(16)-dihydroxy-11-oxo-5Z,13E-prostenoic acid (3,3,4,4-d4)	C <sub>20</sub> H <sub>32</sub> O <sub>5</sub>	356.25
LIPAC0000017	PGE <sub>2</sub> -d4	11R,15(16)-dihydroxy-9-oxo-5Z,13E-prostenoic acid (3,3,4,4-d4)	C <sub>20</sub> H <sub>32</sub> O <sub>5</sub>	356.25
LIPAC0000018	PGF <sub>2</sub> -d4	9,15(16)-epoxy-15(16)-hydroperoxy-5Z,13E-prostenoic acid	C <sub>20</sub> H <sub>32</sub> O <sub>5</sub>	368.22

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LIPID Metabolites And Pathways Strategy

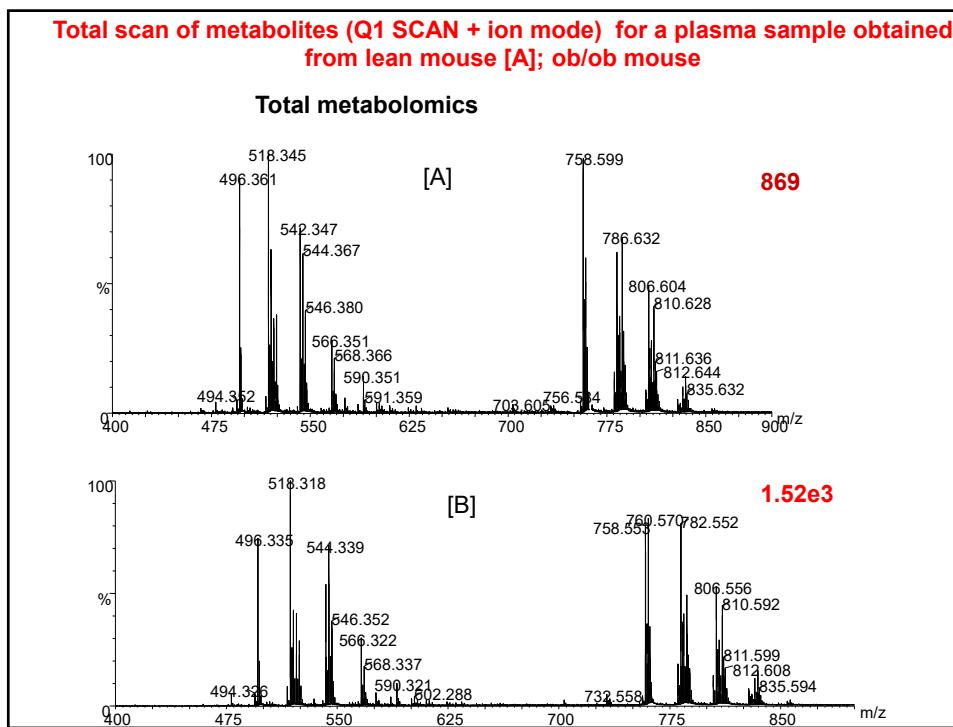
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Structure database (LMSD)

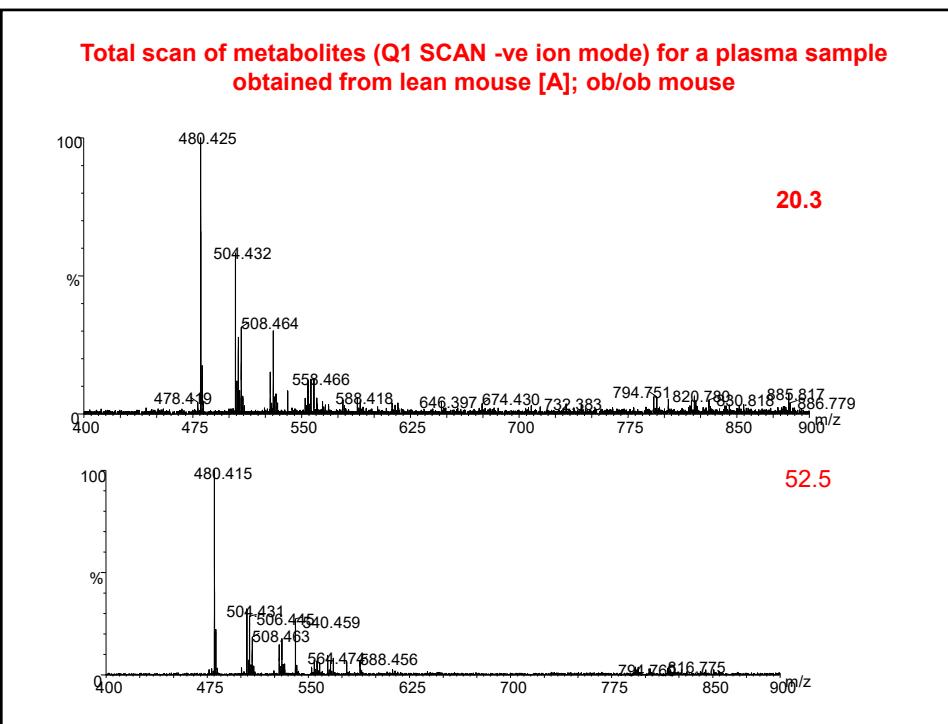
LMFA03010025

LM ID LMFA03010025  
 Common Name PGF $\beta$   
 Systematic Name 9R,11R,15S-trihydroxy-5Z,13E-prostaglandin F<sub>2</sub>  
 Synonyms -  
 Exact Mass 354.24  
 Formula C<sub>20</sub>H<sub>34</sub>O<sub>5</sub>  
 Category Fatty Acyls [FA]  
 Main Class Eicosanoids [FA03]  
 Sub Class Prostaglandins [FA0301]  
 LIPIDBANK ID XPR1764  
 PubChem Substance ID 4265968  
 (SID)  
 KEGG ID -

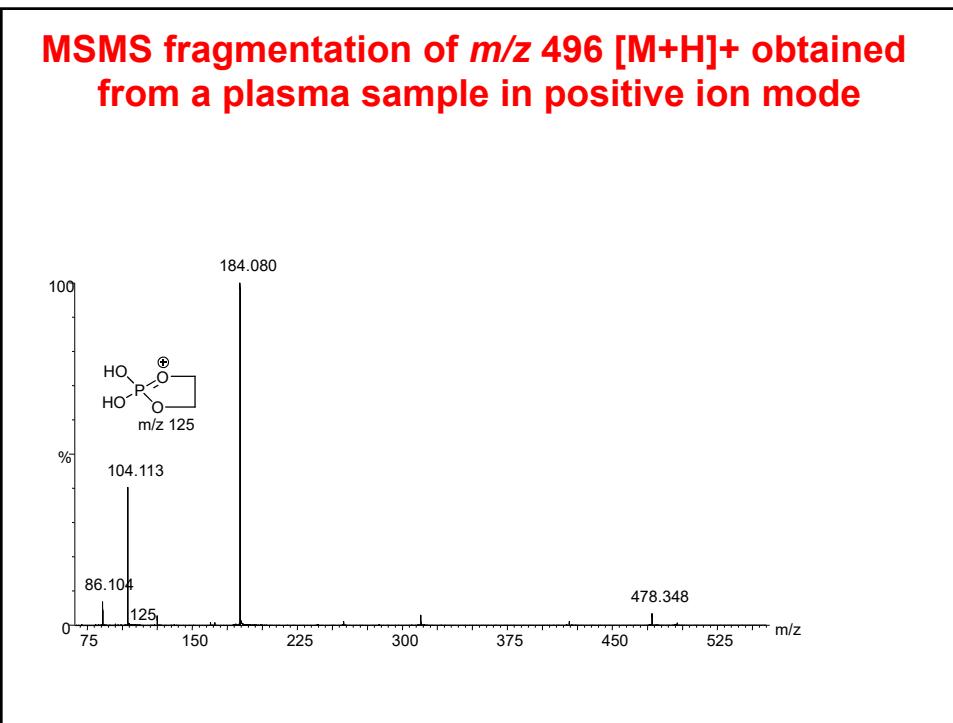
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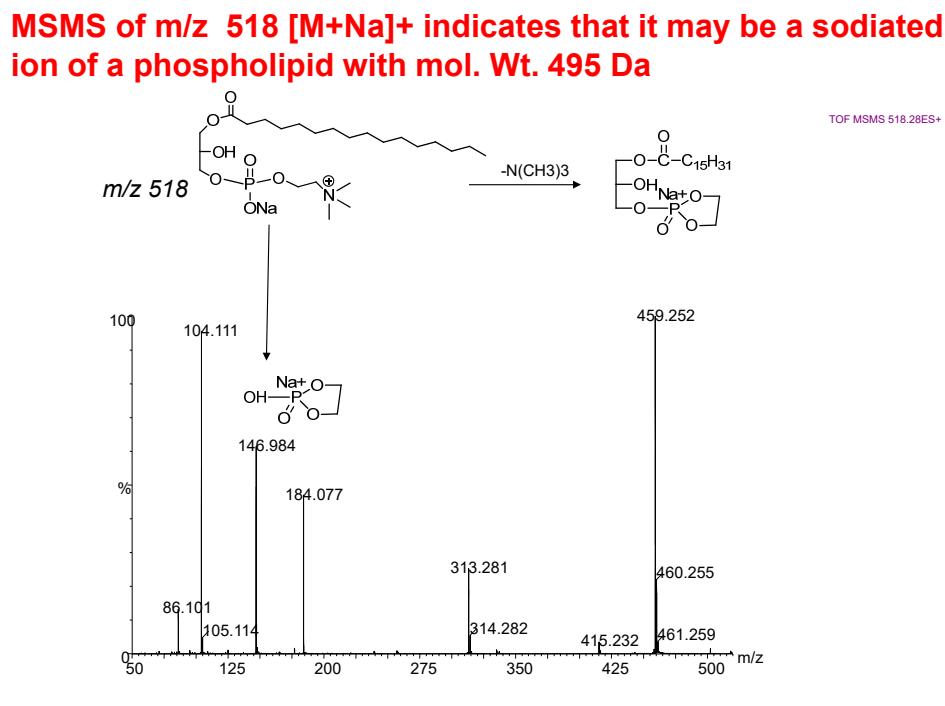
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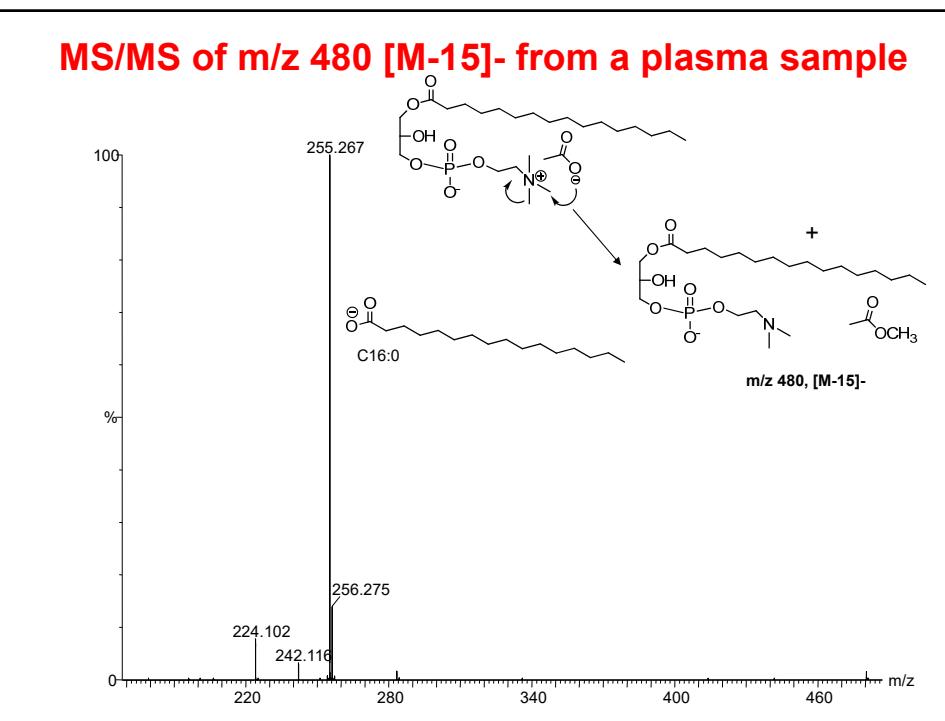
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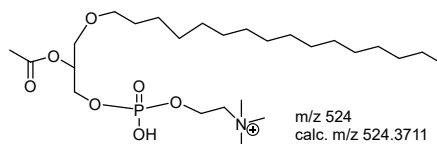
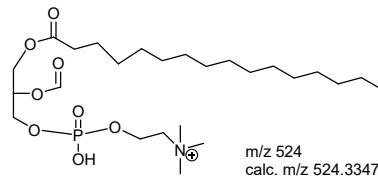
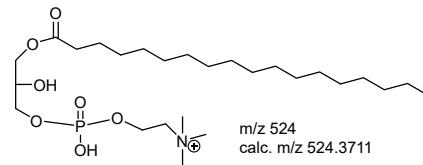


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**Several isobaric compounds-  
Identification by high resolution mass spectrometry**



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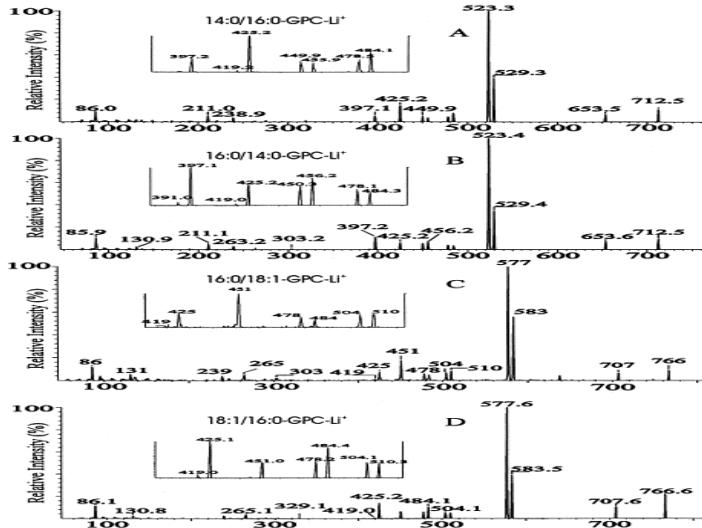
**Lithiated adducts of phosphocholine provide more structural information in their MS/MS spectra**



Source: Hsu et al. J. Am Soc. Mass Spectrom, 1998

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**Relative abundances of product ion can be used to distinguish positional isomers of lithiated phospholipids**



Source: Hsu et al. J. Am Soc. Mass Spectrom., 1998

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## Conclusions

- Shotgun lipidomics approaches are high throughput and applicable to perform profiling as well as quantitative analysis of various lipids in biological samples.
- Tandem mass spectrometry analysis of phospholipids in +ve ion mode characterizes phospholipid polar head groups, whereas -ve ion mode provide fatty acid chain structural information.
- Identification of phospholipids at a molecular level present a great challenge due to their structural diversity and dynamic metabolism.

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